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 NEWS 14 OCT 21 BIOSIS file reloaded and enhanced  
 NEWS 15 OCT 28 BIOSIS file segment of TOXCENTER reloaded and enhanced  
 NEWS 16 NOV 24 MSDS-CCOHS file reloaded  
 NEWS 17 DEC 08 CABA reloaded with left truncation  
 NEWS 18 DEC 08 IMS file names changed

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 MACINTOSH VERSION IS V6.0b(ENG) AND V6.0Jb(JP),  
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 DICTIONARY FILE UPDATES: 7 DEC 2003 HIGHEST RN 624286-58-4

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=> s 65996-63-6/rn

L1 1 65996-63-6/RN

=> d l1 all

L1 ANSWER 1 OF 1 REGISTRY COPYRIGHT 2003 ACS on STN

RN 65996-63-6 REGISTRY \*

\* Use of this CAS Registry Number alone as a search term in other STN files may result in incomplete search results. For additional information, enter HELP RN\* at an online arrow prompt (=>).

ED Entered STN: 16 Nov 1984

CN Starch, acid-hydrolyzed (CA INDEX NAME)

DEF The substance resulting from treatment of starch in aqueous slurry with small amounts of an acidifying substance, such as aluminum sulfate, generally at elevated temperature and pressure. The process is usually continuous with the time of treatment very short. Degree of hydrolysis is usually determined by measurement of viscosity and controlled by rate of the slurry through the reactor, temperature, and acidity.

MF Unspecified

CI MAN, GRS

LC STN Files: CHEMLIST, MSDS-OHS, USPATFULL

Other Sources: DSL\*\*, TSCA\*\*

(\*\*Enter CHEMLIST File for up-to-date regulatory information)

**STRUCTURE DIAGRAM IS NOT AVAILABLE**

=> file caplus

COST IN U.S. DOLLARS

SINCE FILE	TOTAL
ENTRY	SESSION
3.33	3.54

FULL ESTIMATED COST

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FILE COVERS 1907 - 9 Dec 2003 VOL 139 ISS 24  
 FILE LAST UPDATED: 8 Dec 2003 (20031208/ED)

This file contains CAS Registry Numbers for easy and accurate substance identification.

=> s 11  
 L2 0 L1

=> del 12 y

=> s 65996-63-6/rn  
 0 65996-63-6  
 0 65996-63-6D  
 L2 0 65996-63-6/RN  
 (65996-63-6 (NOTL) 65996-63-6D )

=> s 65996-63-6#/rn  
 L3 0 65996-63-6#/RN  
 (65996-63-6#)

=> s acid (P) hydrolyzed (P) starch  
 3753608 ACID  
 1414495 ACIDS  
 4215859 ACID  
 (ACID OR ACIDS)  
 131932 HYDROLYZED  
 1 HYDROLYZEDS  
 131933 HYDROLYZED  
 (HYDROLYZED OR HYDROLYZEDS)  
 137076 STARCH  
 8076 STARCHES  
 137970 STARCH  
 (STARCH OR STARCHES)  
 L4 1516 ACID (P) HYDROLYZED (P) STARCH

=> s 14 and coal  
 205824 COAL  
 35049 COALS  
 207632 COAL  
 (COAL OR COALS)  
 L5 8 L4 AND COAL

=> d 15 1-8 all

L5 ANSWER 1 OF 8 CAPLUS COPYRIGHT 2003 ACS on STN

Full Text	Citing References
AN 1999:219773 CAPLUS	
DN 130:256814	
ED Entered STN: 08 Apr 1999	
TI Starch/cationic polymer combinations as coagulants for the mining industry	
IN Jankowski, Jeffrey A.; Tobison, Calvin T.	
PA Nalco Chemical Company, USA	
SO Eur. Pat. Appl., 22 pp.	
CODEN: EPXXDW	
DT Patent	
LA English	
IC ICM C02F011-14	
ICS C02F001-56	
CC 60-2 (Waste Treatment and Disposal)	
Section cross-reference(s): 51, 54	

FAN.CNT 2

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
EP 905091	A1	19990331	EP 1998-307846	19980928
R: AT, BE, CH, DE, DK, ES, FR, GB, GR, IT, LI, LU, NL, SE, MC, PT, IE, SI, LT, LV, FI, RO				
CA 2248479	AA	19990329	CA 1998-2248479	19980928

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AU 9887153	A1	19990415	AU 1998-87153	19980929
AU 739878	B2	20011025		
ZA 9808880	A	19990428	ZA 1998-8880	19980929
PRAI US 1997-939249	A	19970929		

AB A method for dewatering **coal** tailings, clean **coal** products and mineral slurries with an effective coagulating amt. of a combination of a cationic polymer and a starch. A preferred cationic polymer is poly(dimethylaminoethylacrylate Me chloride quaternary salt) and preferred starches are unmodified.

ST **coal** tailings dewatering coagulant; mineral tailings dewatering coagulant

IT Mining

Solid wastes  
(**coal**; starch/cationic polymer combinations as coagulants for mining industry slurries and tailings)

IT Acrylic polymers, uses  
RL: MOA (Modifier or additive use); USES (Uses)  
(starch/cationic polymer combinations as coagulants for mining industry slurries and tailings)

IT Taconite  
RL: PEP (Physical, engineering or chemical process); PROC (Process)  
(starch/cationic polymer combinations as coagulants for mining industry slurries and tailings)

IT Solid wastes  
(tailings; starch/cationic polymer combinations as coagulants for mining industry slurries and tailings)

IT 9004-53-9, Dextrin  
RL: MOA (Modifier or additive use); USES (Uses)  
(1719 dextrin; starch/cationic polymer combinations as coagulants for mining industry slurries and tailings)

IT 9005-25-8, Corn Starch, uses  
RL: MOA (Modifier or additive use); USES (Uses)  
(C-Gel, Drillstar EW, Min-Star 2050, Pearl Starch, Starpol 410; starch/cationic polymer combinations as coagulants for mining industry slurries and tailings)

IT 9050-36-6, Maltodextrin  
RL: MOA (Modifier or additive use); USES (Uses)  
(Star Dri 100; starch/cationic polymer combinations as coagulants for mining industry slurries and tailings)

IT 221456-85-5, Starmic 620  
RL: MOA (Modifier or additive use); USES (Uses)  
(Starmic 620; starch/cationic polymer combinations as coagulants for mining industry slurries and tailings)

IT 9037-22-3D, Amylopectin, **acid hydrolyzed**  
RL: MOA (Modifier or additive use); USES (Uses)  
(pregelatinized; **starch**/cationic polymer combinations as coagulants for mining industry slurries and tailings)

IT 26062-79-3 54076-97-0 221548-92-1, Nadex 772 221549-07-1, X-Pand'r  
RL: MOA (Modifier or additive use); USES (Uses)  
(starch/cationic polymer combinations as coagulants for mining industry slurries and tailings)

IT 7440-41-7, Beryllium, processes 7440-50-8, Copper, processes  
13463-67-7, Titania, processes 15243-87-5, Trona  
RL: PEP (Physical, engineering or chemical process); PROC (Process)  
(starch/cationic polymer combinations as coagulants for mining industry slurries and tailings)

RE.CNT 3 THERE ARE 3 CITED REFERENCES AVAILABLE FOR THIS RECORD

RE

- (1) Dow Chemical; GB 1091087 A
- (2) Dow Chemical; EP 0082571 A 1983 CAPLUS
- (3) Phillip, S; US 3541009 A 1970 CAPLUS

L5 ANSWER 2 OF 8 CAPLUS COPYRIGHT 2003 ACS on STN

Full Citing  
Text References

AN 1989:410127 CAPLUS  
 DN 111:10127  
 ED Entered STN: 08 Jul 1989  
 TI Adsorbents for drying of **coal** powder  
 IN Shibano, Takeshi; Fujimoto, Takashi; Kato, Koji  
 PA Mitsubishi Petrochemical Co., Ltd., Japan  
 SO Jpn. Kokai Tokkyo Koho, 4 pp.  
 CODEN: JKXXAF  
 DT Patent  
 LA Japanese  
 IC ICM C02F011-14  
 ICS C08L101-00  
 CC 51-17 (Fossil Fuels, Derivatives, and Related Products)  
 FAN.CNT 1

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	JP 63241000	A2	19881006	JP 1987-74138	19870330
PRAI	JP 1987-74138		19870330		

AB Water-contg. **coal** or other mineral powder are dried by contacting with a highly water-adsorbable resin (av. particle diam  $\geq 500 \mu\text{m}$ , preferably 1000-5000  $\mu\text{m}$ , and the spent adsorbent can be sepd. and regenerated by desorption with a water-sol. org. solvent. The water-adsorbable resin includes vinyl alc.-acrylic **acid** copolymer, crosslinked polyvinyl alc., **hydrolyzed starch**-acrylonitrile graft polymer, and crosslinked polyacrylate salts (I). Thus, 100 wt. parts **coal** powder contg. 36% water and 4 wt. parts I (i.e., Diawet) were blended for ~30 min, and then screened to sep. a dried **coal** contg. 5.5% water; the spent resin was regenerated by desorption with a ~45 vol.% aq. acetone soln.

ST **coal** drying adsorbent crosslinked polyacrylate; polyvinyl alc adsorbent  
**coal** drying

IT **Coal** treatment  
 (drying, highly water-adsorbable resin as adsorbent for, for reduced energy cost)

IT 9002-89-5D, Poly(vinyl alcohol), crosslinked 9003-01-4D, Polyacrylic acid, salts, crosslinked 26299-60-5, Vinyl alcohol-acrylic acid copolymer 107830-79-5D, hydrolyzed

RL: USES (Uses)  
 (adsorbent, for drying of **coal** powder)

L5 ANSWER 3 OF 8 CAPLUS COPYRIGHT 2003 ACS on STN

Full Text	Citing References
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AN 1981:474807 CAPLUS

DN 95:74807

ED Entered STN: 12 May 1984

TI Effects of drilling fluids on soils and plants: I. Individual fluid components

AU Miller, Raymond W.; Honarvar, Shahnaz; Hunsaker, Barbara

CS Utah Agric. Exp. Stn., Logan, UT, 84322, USA

SO Journal of Environmental Quality (1980), 9(4), 547-52

CODEN: JEVQAA; ISSN: 0047-2425

DT Journal

LA English

CC 4-3 (Toxicology)

AB The effects of 31 drilling fluid (drilling mud) components on the growth of green beans (*Phaseolus vulgaris*) and sweet corn (*Zea mays* var. *Saccharata*) were evaluated in greenhouse studies. Plants grew well in fertile Dagor silt loam soil (Cumulic Haploxeroll) when the soil was mixed with most soil-component mixts. at disposal proportions normally expected. Vinyl acetate and maleic **acid** polymer addn. caused significantly increased growth at the 95% confidence level. No statistically significant depression of plant growth occurred at normal rates with asbestos, asphalt, barite, bentonite, Ca lignosulfonate [8061-52-7], Na

polyacrylate [9003-04-7], a modified tannin, ethoxylated nonylphenol, a filming amine, gilsonite, a Xanthan gum, paraformaldehyde [30525-89-4], a pipe dope, **hydrolyzed polyacrylamide**, Na **acid** pyrophosphate, Na CM-cellulose [9004-32-4], NaOH added as pellets, and a sulfonated tall oil. Statistically significant redns. in plant yields (at the 95% confidence level) occurred at normal disposal rates with a long-chained aliph. alc., NaCr2O7, diesel oil, guar gum, an Fe chromelignosulfonate, lignite, a modified asphalt, a plant fiber-synthetic fiber mixt., lignite, a nonfermenting **starch**, KCl, pregelatinized **starch**, and sulfated triglyceride. Thirteen drilling fluid components added individually to a fluid base (water, bentonite, and barite) and then to soil, were also tested for their effect on plant growth. Only the sulfated triglyceride (Torq-Trim [72027-04-4]) and the long-chain (high-mol.-wt.) alc. (Drillaid 405 [78413-46-4]) caused no plant growth redns. at either rate added. The modified tannin (Desco [54847-47-1]) caused minimal redn. in bean growth only when added to soil in excess levels. The pregelatinized **starch**, Na2Cr2O7, nonfermenting **starch** (Dextrid [9004-53-9]), plant and synthetic fiber mix (Kwik-Seal [70226-02-7]), Fe chromelignosulfonate (Q-Broxin [8075-74-9]), and guar gum (Gendril Thik [9000-30-0]) did not depress plant growth at normal disposal rates used, but they did depress plant growth with the abnormally high (excess) rates of addn. to soil. No plants grew at either normal or excessive addn. rates when KCl was used. Also, no plants grew when abnormally high rates of Na2Cr2O7, lignite + NaOH, or Ca lignosulfonate + NaOH were added.

ST drilling fluid toxicity bean corn

IT Soil pollution

(by drilling fluids, beans and corn in relation to)

IT Bean

Corn

(drilling fluid components toxicity to)

IT Toxicity

(of drilling fluid components, to beans and corn)

IT Tall oil

RL: BIOL (Biological study)

(sulfonated, toxicity of, to beans and corn, drilling fluids in relation to)

IT Drilling fluids and muds

(toxicity of, to beans and corn)

IT Fuels, diesel

(toxicity of, to beans and corn, drilling fluids in relation to)

IT Asbestos

Asphalt

Bentonite, biological studies

Coal, brown

Gilsonite

RL: ADV (Adverse effect, including toxicity); BIOL (Biological study)

(toxicity of, to beans and corn, drilling fluids in relation to)

IT 1310-73-2, biological studies 7447-40-7, biological studies 7758-16-9  
8061-52-7 8075-74-9 9000-30-0 9003-04-7 9004-32-4 9004-53-9  
9016-45-9 10588-01-9 11138-66-2 13462-86-7 24980-59-4 30525-89-4  
37224-28-5 54847-47-1 70226-02-7 72027-04-4 78413-46-4

RL: ADV (Adverse effect, including toxicity); BIOL (Biological study)

(toxicity of, to beans and corn, drilling fluids in relation to)

L5 ANSWER 4 OF 8 CAPLUS COPYRIGHT 2003 ACS on STN

Full Text	Citing References
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AN 1980:453341 CAPLUS

DN 93:53341

ED Entered STN: 12 May 1984

TI Destabilization of sludges with hydrolyzed starch flocculants

IN Yong, Raymond Nenyiu; Sethi, Amar Jit

PA Great Canadian Oil Sands Ltd., Can.

SO Brit. UK Pat. Appl., 12 pp.

eb

h eb c g cg b cg

CODEN: BAXXDU

DT Patent

LA English

IC C02F001-56; C08L003-02

CC 60-2 (Sewage and Wastes)

FAN.CNT 2

	PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
PI	GB 2027684	A	19800227	GB 1979-27035	19790802
	GB 2027684	B2	19830330		
	CA 1121555	A1	19820413	CA 1978-308619	19780802
	IN 153565	A	19840728	IN 1979-CA789	19790730
	AU 7949406	A1	19800207	AU 1979-49406	19790731
	AU 535348	B2	19840315		
	NL 7905919	A	19800205	NL 1979-5919	19790801
	DE 2931278	A1	19800228	DE 1979-2931278	19790801
	DE 2931278	C2	19890518		
	JP 55061904	A2	19800510	JP 1979-97410	19790801
	JP 62006876	B4	19870213		
	DE 2954628	C2	19901206	DE 1979-2954628	19790801
	IN 153622	A	19840728	IN 1982-CA537	19820513
PRAI	CA 1978-308619		19780802		
	IN 1979-CA789		19790730		

AB Phosphate slimes or aq. colloidal sludge suspensions from **coal** and tar sands mining, contg. clay minerals or metal oxides and/or hydroxides, were destabilized by treatment with wheat, corn, or potato **starch** [9005-25-8] ( $\geq 50$  ppm, **hydrolyzed** in the presence of alkali metal salts), cement ( $\geq 3$  lb/100 Imperial gel bituminous tar sands sludge contg. 20% solids), and a lower aliph. alc., acetone, yeast, or lactic **acid**. E.g., centrifugal and free sedimentation rates were increased by treatment of 50 mL tar sand sludge (10% solids content, 0.25% bitumen) with Ca Al phosphate **hydrolyzed** wheat **starch** (200 ppm) and yeast (80 ppm), alc. (1000 ppm), or lactic **acid** (88 ppm).

ST sludge destabilization hydrolyzed starch

IT Flocculating agents  
(hydrolyzed starch, for sludge destabilization)

IT Waste solids  
(Sludges, destabilization of, with hydrolyzed starch flocculants)

IT 9005-25-8D, hydrolyzed  
RL: PROC (Process)  
(as flocculant for sludge destabilization)

L5 ANSWER 5 OF 8 CAPLUS COPYRIGHT 2003 ACS on STN

Full Text	Citing References
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AN 1969:473801 CAPLUS

DN 71:73801

ED Entered STN: 12 May 1984

TI Treatment of water and sludge

AU Hollo, Janos

CS Tech. Univ., Budapest, Hung.

SO Australian Chemical Engineering (1969), 10(5), 19-24

CODEN: ASCEAE; ISSN: 0004-8828

DT Journal

LA English

CC 60 (Sewage and Wastes)

AB Effectiveness of various polymeric flocculating agents in the treatment of waste water and sludge is discussed. Polyacrylamide (I) agents produced greater sedimentation rates than poly(methacrylic **acid**), poly(acrylic **acid**), **hydrolyzed starch**, or CM-cellulose. Successful sedimentation by these polyelectrolytes was possible only in a well-defined range of solid matter concn. that had both max. and min. concn. limits. Uses of I in the sepn. and dehydration of fine sludge from **coal**-ore dressing plants, for the purification of surface waters, and for the treatment of

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ST industrial sewages were described.  
 polymers flocculation waste waters; polyacrylamides flocculation waste  
 waters; polyelectrolytes flocculation waste waters; flocculation waste  
 waters polyelectrolytes; sedimentation waste waters polyelectrolytes;  
 surface waters purifn polyelectrolytes

IT **Coal**

RL: PROC (Process)  
 (cleaning of, polyacrylamide in flocculation of sludge from)

IT **Wastes**

(coagulation of, by polyelectrolytes)

IT **Water purification**

(coagulation, by polyelectrolytes)

IT **Polyelectrolytes**

(flocculation by)

IT **Cellulose, carboxymethyl ether**

Methacrylic acid, uses and miscellaneous

RL: PROC (Process)

(flocculation by)

IT 9003-01-4 9003-05-8

RL: PROC (Process)

(flocculation by)

IT 9005-25-8, uses and miscellaneous

RL: USES (Uses)

(hydrolyzate, flocculation by)

L5 ANSWER 6 OF 8 CAPLUS COPYRIGHT 2003 ACS on STN

Full Text	Citing References
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AN 1925:11852 CAPLUS

DN 19:11852

OREF 19:1586c-e

ED Entered STN: 16 Dec 2001

TI Sublimation of "unsublimable" substances

AU Kurschner, Karl

SO Mikrochem. (1925), 3, 1-20

DT Journal

LA Unavailable

CC 11D (Biological Chemistry: Botany)

AB K. has subjected the lignin from pine, oak, rye straw, and red beech, prep'd. in various ways, pine shavings, pine mold, sulfite liquors, lignite, brown **coal** and an inclusion in hard **coal** to sublimation and in every case has obtained crystals of vanillic **acid** sometimes mixed with vanillin and NH<sub>4</sub>Cl. The vanillic **acid** is the result of the oxidation of vanillin. The residue resembles caramelized sugar and has a vigorous reducing action on Fehling soln. Because of the formation of this residue together with some HCHO, K. suggests that the vanillic **acid** represents a group that has been combined as a glucoside. Coniferin found in the cambian fluid of plants undergoes the same changes as lignin when sublimed; hence it is probable, according to K., that lignin may be a polymerization product of the former. When the complex is heated the carbohydrate portion becomes dehydrated and caramelized, the coniferyl alcohol is **hydrolyzed** from it and oxidized to vanillin, vanillic **acid** and AcOH. **Starch** and various tannins also yield considerable quantities of sublimates. These have not been investigated.

IT Strawberries

(acids of)

IT Sublimation

(of unsublimable substances)

IT 121-34-6, Vanillic acid

(in lignin)

IT 531-29-3, Coniferin 9005-53-2, Lignin

(sublimation of)

L5 ANSWER 7 OF 8 CAPLUS COPYRIGHT 2003 ACS on STN

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Full Text	Citing References
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AN 1924:6485 CAPLUS  
 DN 18:6485  
 OREF 18:887e-g  
 ED Entered STN: 16 Dec 2001  
 TI Built-up adsorbent charcoal  
 IN Morrell, J. C.  
 DT Patent  
 LA Unavailable  
 CC 18 (Acids, Alkalies, Salts, and Sundries)  
 FAN.CNT 1

PATENT NO.	KIND	DATE	APPLICATION NO.	DATE
US 1478985		19231225	US	

PI Carbon black or the like is freed from volatile oily impurities by  
 AB treatment with steam at a temp. of 450-600° for several hrs., mixed  
 with an emulsified binder which may be formed of pitch, NH<sub>4</sub> tannate,  
 tannic acid, anthracene oil and H<sub>2</sub>O, dried, molded under a pressure of  
 16-30 tons per sq. in. and crushed or ground. The material is then  
 activated by heating (e. g., to 850° for 4 hrs., to 925° for  
 1/2 hr. and to 925-50° for 3 hrs.). A further heating (which may  
 be with accompanying steam treatment) after partial cooling improves the  
 qualities of the product and a third heating also effects further  
 improvement. U. S. 1,478,986 relates to a similar process in which the C  
 initially used is mixed with a binder formed of pitch, coal tar, or  
 similar material dissolved in C<sub>6</sub>H<sub>6</sub>, solvent naphtha, toluene, xylene,  
 CCl<sub>4</sub>, a paraffin hydrocarbon oil of low b. p. or a similar volatilizable  
 solvent, before molding and heat-treatment. U. S. 1,478,987 specifies  
 preliminary mixing of the C with a non-fluid binder which chars on  
 heating, e. g., a sugar, starch, pectins, proteins, hydrolyzed wood  
 waste, sulfite-liquor pitch, coal-tar or resin pitch.  
 IT Charcoal  
 (adsorbent)

L5 ANSWER 8 OF 8 CAPLUS COPYRIGHT 2003 ACS on STN

Full Text	Citing References
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AN 1917:796 CAPLUS  
 DN 11:796  
 OREF 11:173i,174a-e  
 ED Entered STN: 16 Dec 2001  
 TI Chemical composition, digestibility, and feeding value of vegetable-ivory  
 meal  
 AU Beals, C. L.; Lindsey, J. B.  
 CS Mass. Agr. Expt. Sta.  
 SO Journal of Agricultural Research (Washington, D. C.) (1916), 7, 301-20  
 CODEN: JAGRAC; ISSN: 0095-9758  
 DT Journal  
 LA Unavailable  
 CC 12 (Foods)  
 AB Vegetable ivory is the seed or nut of Phytelphas macrocarpa, and is used  
 for the manufacture of buttons, etc. The resultant sawdust, chips, and  
 turnings has been mixed with other ingredients to be used as a cattle  
 food. Analyses show vegetable ivory to be carbohydrate in nature,  
 containing about 5% protein and 75% N-free extract. Fat and mineral  
 matter are negligible, while crude fiber averages 7%. 92.5% of the N-free  
 extract is mannan, a polymer of mannose. Pentosans are present to the  
 extent of 2.5%. Lignin, galactan, starch, and dextran were not found.  
 A non-nitrogenous "alcoholic ppt." amounting to about 2.5% is present; it  
 is not pentosic in nature. It differs from fruit "pectin" in that it does  
 not form mucic acid and does not reduce Cu. By the use of Fehling's  
 soln. about 0.5% H<sub>2</sub>O-sol. reducing material and 2% so-called total sugars  
 are shown to be present, after inversion with HCl in the cold. The mannan

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is not entirely **hydrolyzed** without at least 4 1/2 hrs. boiling in an **acid** soln. The characteristic "**acid**" color of the soln. bleaches out at the completion of hydrolysis. Practically the entire N-free extract is accounted for in the form of a hexose sugar or its condensation product, except a small % of pentoses and pectin. The energy equiv. of the material ranks well with other carbohydrate foods, and is equal to 1/2 that of soft **coal**. Sheep ate vegetable-ivory meal readily when it was mixed with other grains. 84% of the dry matter and 92% of the N-free extract were digested. All the carbohydrates appeared to have been **hydrolyzed** and absorbed in the digestive tract. Cows ate the material when mixed with other food, without evidence of digestive disturbances. They refused to eat it if fed by itself. When fed in mixed food, the increase in milk was sufficient to indicate its positive value as a productive feed. Though exact relative values cannot be shown, it seems certain that vegetable-ivory meal does not fully equal corn meal for milk production. A list of 14 citations is appended.

IT Ivory nuts  
(meal)

IT Phosphates  
(recovery of, from natural phosphates)

=> file stnguide

COST IN U.S. DOLLARS

SINCE FILE	TOTAL
ENTRY	SESSION
35.06	38.60

FULL ESTIMATED COST

DISCOUNT AMOUNTS (FOR QUALIFYING ACCOUNTS)

SINCE FILE	TOTAL
ENTRY	SESSION
-5.21	-5.21

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FILE CONTAINS CURRENT INFORMATION.

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